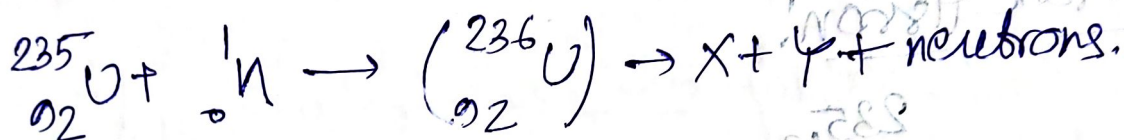


Nuclear Fission

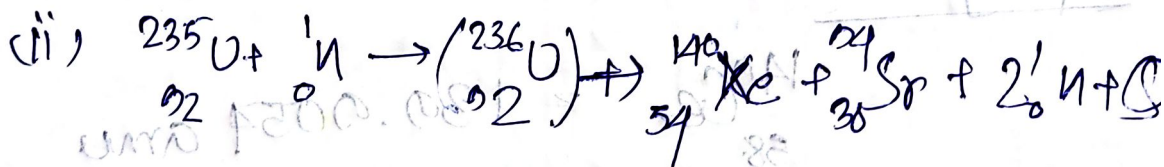
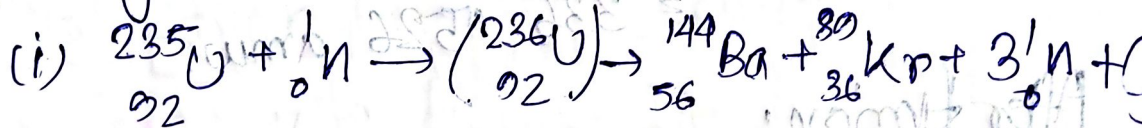
The phenomena of the division of a heavy nucleus into two nearly equal parts is termed as Nuclear fission.

The schematic eqⁿ for the fission process may be written as.



Here X & Y are the fission fragment

Typically fission reaction are



Energy Released in Fission.

The Energy released in the fission process is so high that the two main fission products fly apart in opposite direction with great speeds as compared

by many cloud chamber photographs.

Actual kinetic energy of the separate particle is about 180 MeV. & the energy of the gamma radiation emitted in the process is about 20 MeV.

The following calculation is given below.

Before fission:

$${}_{92}^{235}\text{U} = 235.0439 \text{ amu.}$$

$${}_{0}^1\text{n} = 1.0087 \text{ amu.}$$

$$\text{Total} = 336.0526 \text{ amu.}$$

After fission:

$${}_{58}^{140}\text{Ce} = 139.9054 \text{ amu}$$

$${}_{40}^{94}\text{Zr} = 93.9036 \text{ amu.}$$

$$2 {}_{0}^1\text{n} = 2.0173 \text{ amu.}$$

$$6 \beta^{-} = 0.0033 \text{ amu}$$

$$\text{Total} = 235.8296 \text{ amu}$$

mass difference = $(236.0526 - 235.8206)$ amu,
= 0.223 amu.

This represents a Q-value given by

$$Q = 0.223 \times 931 = 208 \text{ MeV.}$$

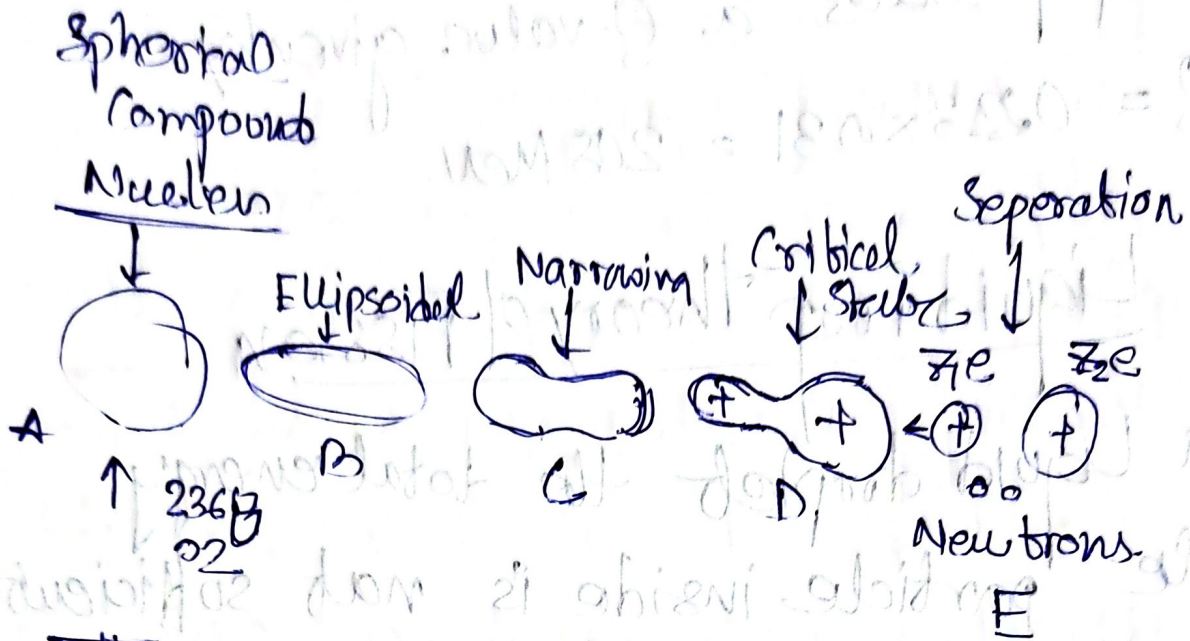
Liquid Drop Theory of Fission

In a liquid droplet the total energy of the particle inside is not sufficient enough to overcome the forces that hold the particle together in the form of droplet. But if some energy is added to the droplet so as to set into vibration, it will alternately elongate & contract. If the amplitude of vibration be sufficiently large the droplet might divide itself into two more or less equal parts.

~~The coulomb r~~

According to Bohr's such a situation arises in the

nucleus when bombarded with neutron & results in fission.



The value of the critical energy,

$$E_{\text{crit}} = 0.80 A^{2/3} - 0.02 \frac{Z(Z-1)}{A^{1/3}} \text{ MeV}$$

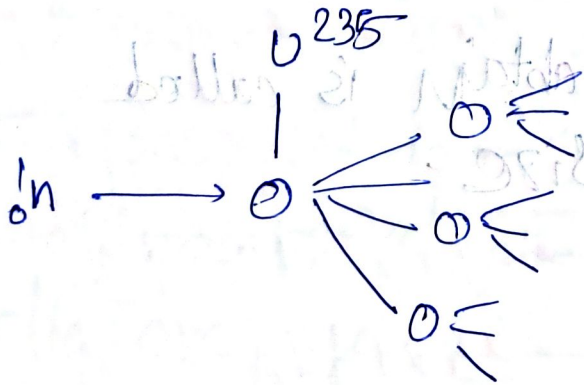
* Critical Energy of $^{235}_{92}\text{U}$

$$A = 235, Z = 92$$

$$\begin{aligned}
 E_{\text{crit}} &= 0.80(235)^{2/3} - 0.02 \left(\frac{92(92-1)}{(235)^{1/3}} \right) \\
 &= 33.9884486 - 27.0049621 \\
 &= \cancel{33.8923681} - \cancel{27.133402} \\
 &= \cancel{6.7599270} \text{ MeV} = 6.8034 \\
 &= \cancel{6.76} \text{ MeV} \approx \underline{6.8 \text{ MeV}}
 \end{aligned}$$

Nuclear Chain Reaction

In each event of a fission reaction with heavy nuclei, the highly excited nuclei, emit two or three prompt neutrons as correctly suggested by Fermi, this fission producing neutrons interacts with neighbouring nuclei & produce further fission in them, so that more neutrons are emitted the result is an avalanche like buildup of fission events. Such a fission reaction is called a Chain Reaction



Critical Mass

The important point in the possibility of achieving of chain reaction is the size of the core. The core is the space in which the chain reaction takes place. Reduction in the size of the core would increase, the number of the neutrons that escape beyond the core & would thus reduce the possibility of further development of chain reaction.

The minimum size of the core at which a chain reaction can still be obtained is called the critical size.

Reproduction Factor

Let, at an instant there be N_0 , ~~first~~ fast neutrons available to produce a fission. A few of this may induce fission in natural U-238, so that the available neutrons increase in number by a factor μ & become $(N_0)\mu$. If small μ or the fraction slowed down, then the available neutrons are $N_0\mu p$ of this again a fraction f may succeed in producing fission in U-235.

Before being lost by diffusion. So the number U-235 is $N_0\mu p f$. If at each fission U-235, c fast neutrons are generated to restart the cycle. The total number of ~~electron~~ neutron after one cycle is

$$N = N_0 \mu p f c$$

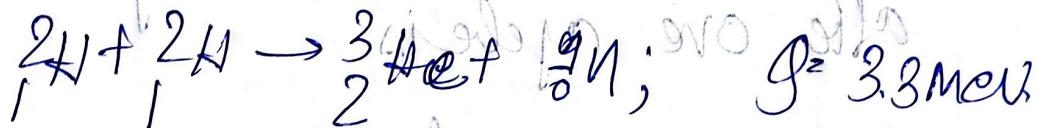
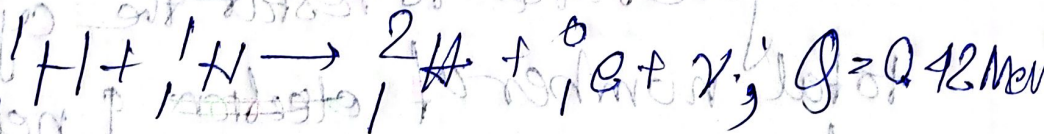
$\frac{N}{N_0} \cdot \text{exp.} = \text{Reproduction Factor} = k$

$k = \frac{\text{Rate of neutron production}}{\text{Rate of neutron disappearance}}$

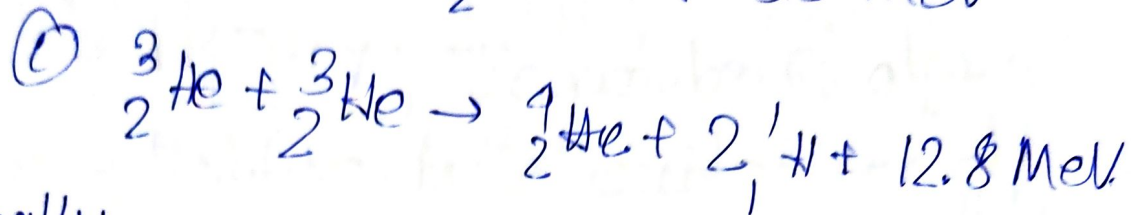
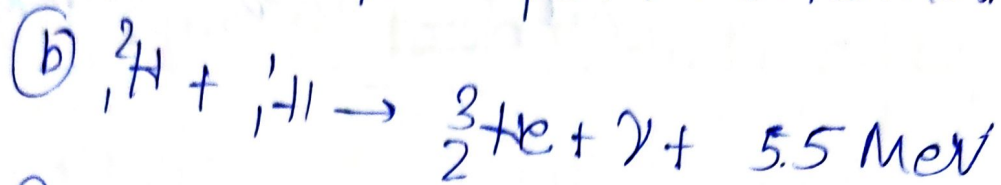
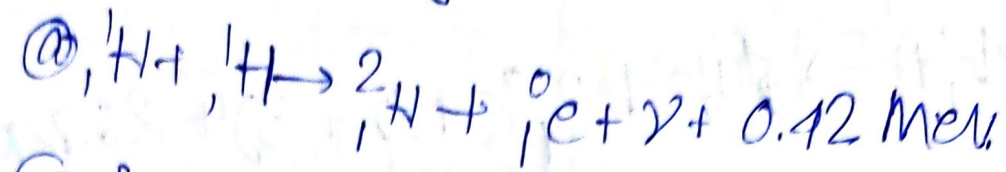
Nuclear Reactions

Nuclear Fusion

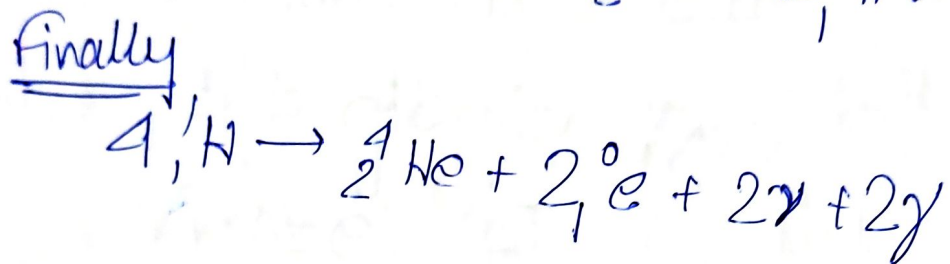
If two light nuclei combined or fused together to produce a relatively heavier nucleus, there would be a greater binding energy and consequent decrease in nuclear mass. This type of nuclear reaction is known as Nuclear Fusion.



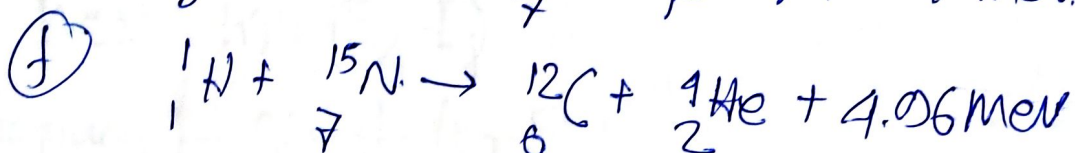
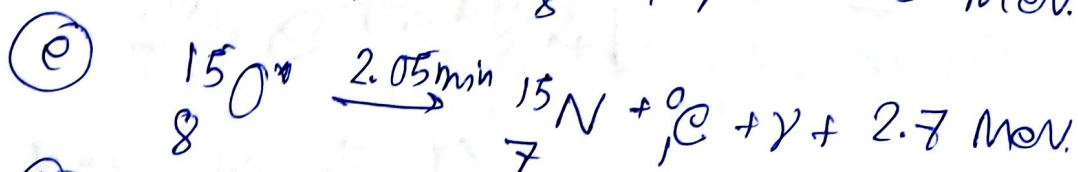
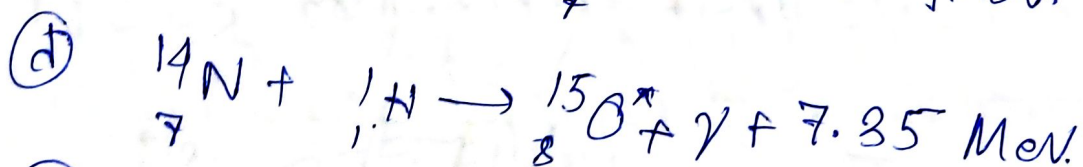
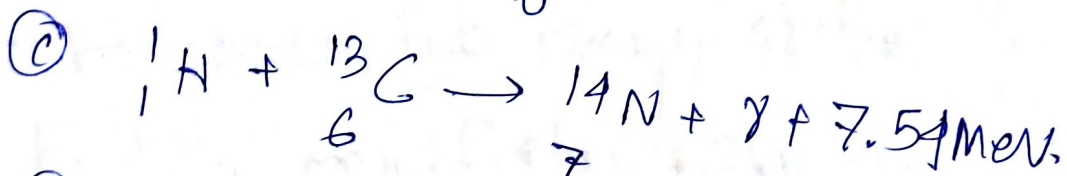
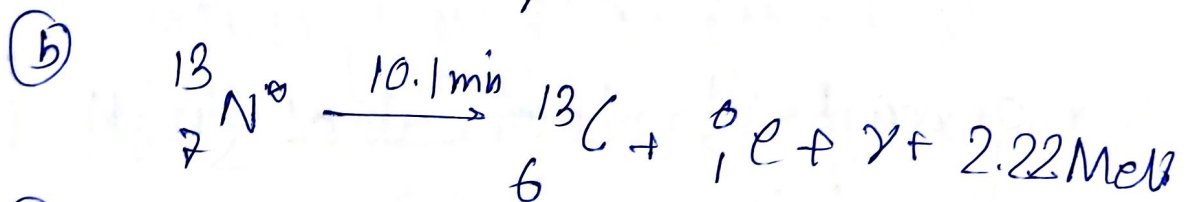
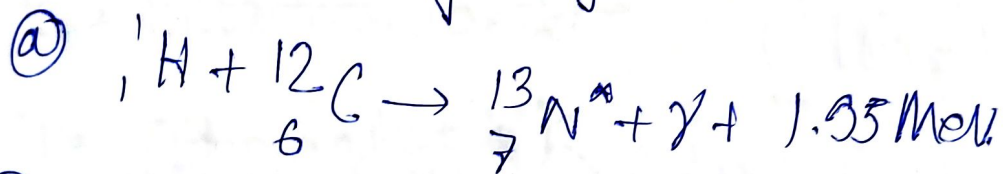
Proton-Proton Cycle



Finally



Carbon-Nitrogen Cycle



The net result is proton-proton

